# **105.** Synthesis of Lipophilic Chemotherapeuticals. Part II. 4-Alkylaminoazobenzene-4'-arsonic Acids.

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In continuation of the programme outlined in Part I (J., 1939, 1) we have studied the effect on the biological properties of 4-aminoazobenzene-4'-arsonic acid of introducing alkyl groups of increasing molecular weight into the amino-group.

A SERIES of dyes,  $R\cdot NH\cdot C_6H_4\cdot N:N\cdot C_6H_4\cdot AsO_3H_2$ , was prepared by coupling diazotised *p*-arsanilic acid with a solution of the substituted aniline, usually in glacial acetic acid. The lower members of the series are very toxic, the higher ones where  $R = C_{12}H_{25}$ ,  $C_{14}H_{29}$ ,  $C_{16}H_{33}$ , and  $C_{18}H_{37}$  show a definite decrease in toxicity. The methyl compound, *e.g.*, is lethal for mice in doses of 0.05 g. per kg. (corresponding to 0.011 g. of arsenic per kg.), whereas a dose of the octadecyl compound of 0.8 g. per kg. is tolerated (0.105 g. of arsenic per kg.). The benzyl compound is about as toxic as the methyl derivative, 0.05 g. (0.0091 g. of arsenic) per kg. being lethal. No decrease in toxicity was observed when the 4aminophenyl residue—in the 2-ethylhexyl compound ( $CH_3\cdot CH_2\cdot CH$ 

#### EXPERIMENTAL.

Alkylanilines.—The methyl, dimethyl, ethyl, and benzyl compounds were commercial samples; the others were prepared as follows: Aniline (3 mols.) and the appropriate alkyl bromide (1 mol.) were heated for 3 hours on the water-bath, the mass poured into alkali solution, the organic layer washed with water, and the alkylaniline isolated by fractional vacuum distillation. The new members of the series are listed in Table I; the *n*-amyl, *iso*amyl, *n*-hexyl, and *n*-heptyl compounds have been described by Hickinbottom (J., 1932, 2396; 1937, 1119), and the cholesteryl compound by Lieb *et al.* (Annalen, 1935, 509, 222).

### TABLE I.

### Alkylanilines, $R \cdot NH \cdot C_{6}H_{5}$ .

				Foun	d, %.	Requir	ed, %.
R.	В. р.	$n_{\rm D}^{26^{\circ}}$ .	Formula.	С.	H.	Ĉ.	Н.
secButyl	225°/759 mm.; 112-114°/22 mm.	1.5318	$C_{10}H_{15}N$	79.7	10.3	80.5	10.0
secButylcarbinyl	236°/758 mm.; 142°/25 mm.		$C_{11}H_{17}N$	81.5	$9 \cdot 8$	<b>81</b> ·0	10.4
2-Methylpentyl	138°/22 mm.	1.5241	$C_{12}H_{19}N$	81·2	10.7	<b>81·3</b>	10.7
Dodecyl	140°/0·2 mm.; m. p. 28°		$C_{18}H_{31}N$	82.5	11.3	82.7	11.9
Tetradecyl	180°/4 mm.; m. p. 42° *		$C_{20}H_{35}N$	$82 \cdot 4$	12.0	<b>83</b> ·0	12.1
Octadecyl	196°/0·6 mm.; m. p. 42°		$C_{24}H_{43}N$	<b>83</b> ∙6	$12 \cdot 2$	83.5	12.5

\* Crystallised from methanol. Myristylaniline was characterised as 3:5-dinitrobenzomyristylanilide (from light petroleum), m. p. 83°.

4-Alkylaminoazobenzene-4'-arsonic Acids.—A solution of atoxyl (11.9 g.) in a mixture of concentrated sulphuric acid (4.8 c.c.) and water (125 c.c.) was diazotised with sodium nitrite (3.5 g.) and added to the equivalent amount of the substituted aniline, dissolved in glacial acetic acid (four times its weight). The product was collected after 24 hours' standing at room temperature and purified by trituration with and crystallisation from a suitable solvent. The methyl compound has been prepared by Jacobs and Heidelberger (J. Amer. Chem. Soc., 1921, 43, 1632). Most of the dyes had no definite m. p., but they all crystallised well (Table II).

Toxicity Tests.—These were carried out with white mice (Table III). Only the butyl and the sec.-butyl compound have been tested on rats.

The Trypanocidal Properties of 4-Octadecylaminoazobenzene-4'-arsonic Acid (W 47).—A solution of the acid in the equivalent amount of aqueous sodium hydroxide was injected intraperitoneally. In Table IV its activity is compared with that of the minimum curative doses of atoxyl.

4-A	lkylaminoazobenz	zene-4'	-arsonic ac	ids, R•	NH•C <sub>6</sub> H <sub>4</sub> •1	$\mathbf{N} \cdot \mathbf{C}_{6} \mathbf{H}_{4} \cdot \mathbf{A} \mathbf{s} \mathbf{O}_{3} \mathbf{H}_{2}.$
			d. %.	Required %		
R.	Formula.	Ċ.	́н.	Ċ.	Н.	Remarks.
[Methy]	C., H., O.N.Asl	<b>46</b> ·3	4.1	<b>46</b> .6	4.2	<i>i</i>
Dimethyl	$C_{14}H_{16}O_3N_3As$	48.7	$\overline{4} \cdot \overline{3}$	<b>4</b> 8·1	<b>4</b> ·6	From aqueous formic acid; m. p. 310° (decomp.)
Ethyl	$\mathrm{C}_{14}\mathrm{H}_{16}\mathrm{O}_{3}\mathrm{N}_{3}\mathrm{As}$	<b>48</b> ∙2	$5 \cdot 0$	<b>48</b> ·1	<b>4</b> ·6	From aqueous alcohol; m. p. 276° (decomp.)
n-Propyl	$\mathrm{C_{15}H_{18}O_{3}N_{3}As}$	<b>48</b> ·9	5.3	<b>49</b> ·6	$5 \cdot 0$	From glacial acetic acid; m. p. 286° (decomp.)
n-Butvl	C16HaaOaNaAs	50.4	5.5	51.0	5.3	From alcohol
isoButvl	C, H.O.N.As	50.4	5.3	51.0	5.3	From 70% methanol; brick-
	10 20 3 3		(N, 11.8)		(N, 11·1)	red; m. p. 303° (decomp.)
secButyl	$\mathrm{C_{16}H_{20}O_{3}N_{3}As}$	50.8	5.0	<b>51</b> ·0	5.3	From alcohol; contains I mol. of alcohol of crystallisation
n-Amyl	$C_{17}H_{22}O_3N_3As$	$52 \cdot 1$	5.1	$52 \cdot 2$	5.6	From aqueous alcohol or glacial acetic acid
isoAmyl	$\mathrm{C_{17}H_{22}O_3N_3As}$	<b>53</b> ·0	$5 \cdot 0$	52·2 •	5.6	From 80% alcohol; resembles chromic acid in appearance
secButylcarbinyl	$\mathrm{C_{17}H_{22}O_3N_3As}$	51.8	5.6	$52 \cdot 2$	$5 \cdot 6$	Needles from glacial acetic acid; m. p. 245° (decomp.)
n-Hexyl	$\mathrm{C_{18}H_{24}O_{3}N_{3}As}$	53.0	6.8	53.3	$5 \cdot 9$	From 50% acetic acid; m. p. 270° (decomp.)
2-Methylbentyl	C., H., O.N.As	$52 \cdot 8$	5.5	53.3	5.9	M. p. 265° (decomp.)
n-Heptvl	C.H.O.N.As	55.2	6.5	54·4	6.2	From alcohol
n-Dodecvl	C.H.O.N.As	60.0	7.3	58.9	7.4	From 80% acetic acid, brick-red
n-Tetradecvl	C.H.O.N.As	61.5	$8 \cdot 2$	60.3	7.7	From alcohol; brown-red plates
n-Octadecvl	C, H, O, N, As	62.3	8.1	$62 \cdot 9$	8.4	From alcohol; orange leaflets
cycloHexyl	C <sub>18</sub> H <sub>22</sub> O <sub>3</sub> N <sub>3</sub> As	53.7	5.7	53.6	5.4	From glacial acetic acid or formic acid; brown-red; m. p 292° (decomp.)
Benzyl	$\mathrm{C_{19}H_{18}O_3N_3As}$	55.4	4·2	55.5	4.4	From glacial acetic acid and alcohol; * brown; m. p. 340° (decomp.)
Cholesteryl	C <sub>39</sub> H <sub>56</sub> O <sub>3</sub> N <sub>3</sub> As	67.3	8.3	67.9	8.1	From glacial acetic acid and alcohol; † dark red; m. p. 237° (decomp.)

\* Can be purified through the sodium salt, which is precipitated from sodium carbonate solution in brown crystals.

† The dye is soluble in benzenc.

### TABLE III.

### Toxicity tests.

R.	Max. dose tolerated (g. per kg.).	Min. lethal dose (g. per kg.).	R.	Max. dose tolerated (g. per kg.).	Min. lethal dose (g. per kg.).
н	0.01	0.02	2-Methylpentyl	0.02	0.04
Methyl	0.025	0.05	Heptyl	0.05	0.1
Dimethyl	0.04	0.05	2-Ethylhexyl	0.04	0.05
Ethyl	0.01	0.02	Dodecyl	0.3	0.4
Propyl	0.02	0.04	Tetradecyl	0.5	0.6
Butvl	0.01	0.02	Hexadecyl (cetyl)	0.7	0.8
isoButyl	0.02	0.04	Octadecyl	0.8	0.9
secButyl	0.01	0.02	Benzyl	0.025	0.02
n-Amyl	0.01	0.02	cycloHexyl	0.01	0.02
isoAmyl	0.05	0.075	I-(4'-Arsonobenzeneazo)-		
secButylcarbinyl	0.03	0.04	$2-(\beta-\text{ethylhexylamino})-$		
n-Hexyl	0.02	0.1	naphthalene	0.04	0.05

## TABLE IV.

W	47,	dose (g. per kg.).	Trypanosoma brucei.*	Trypanosoma equiperdum.
		0.4	No effect	Effective; relapse after 9 days
		0.5	Effective, but relapse after about 15 days	
		0.6		Effective; relapse after 11 days
		0.7	Effective, but relapse after 30 days	Effective; relapse after 18 days
		atoxyl	0.2 g.; cure	0.15 g.; cure

\* This strain, which was obtained from the Liverpool School of Tropical Medicine, is sensitive to arsenicals.

### TABLE II.

With W 47 the parasites disappear from the blood only after a relatively long time. The trypanosomes disappear within 12 hours after injection of atoxyl, whereas the above azo-dye gives the same effect only after 48—96 hours. We ascribe this fact to the low reducibility of the octadecylamino-compound, as compared with that of atoxyl (compare Breyer, *Biochem. Z.*, 1939, 301, 65). Therefore the inactivity of small doses is probably due to the active form of the dye not reaching the required concentration in time, and it may be expected that these in themselves inactive doses would produce an effect when combined with subcurative doses of a "normal," *i.e.*, quickly reducible, arsenical, such as atoxyl. Table V shows that this conclusion is justified. In the experiments listed, 4-octadecylaminoazobenzene-4'-arsonic acid (W 47) was combined with N-dichloroacetylarsanilic acid, a new arsenical to be described in a forthcoming paper (W 129), which is active against *Trypanosoma brucei (equiperdum)* in doses of 0·1 (0·05) g. per kg.

#### TABLE V.

	Substance.	Dose (g. per kg.).	Trypanosoma brucei.
w	129	0.025	Relapse after 3 days
W	47	0.2	No action
W	47	0.3	No action
W	47	0.4	Prolonged duration of the illness (by about 2 days)
W	129 + W 47	$0.025 \pm 0.2$	Relapse after 7 days
W	129 + W 47	$0.025 \pm 0.3$	Relapse after 10 days
W	129 + W 47	$0.025 \pm 0.4$	Relapse after 21 days
		• • • • •	Trypanosoma equiperdum.
W	/ 129	0.01	Relapse after 2 days
W	47	0.2	No action
W	129+W 47	0.01 + 0.5	Relapse after 8 days
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